Nam	e:		Date:
	IN	TRODUCTION TO POLYME	CRS
Part	A: Definitions and Examp	bles	
Go to	o the website: <u>https://www.t</u>	houghtco.com/definition-of-poly	mer-605912 and answer the
quesi	tions below:		
1. E	Define <i>polymer</i>		
2. L	ist three properties of polyn	ners and explain these properties	S
3. 0	Give three examples of biopo	olymers and three examples of the	neir functions
4. 0	Give three examples of synth	netic polymers and their applicat	ions
	Complete the table below that	at compares and contrasts thermo	oset plastics with thermoplastic
		Thermoset	Thermoplastic
	Rigidity		
	Reaction to heating		
	Bonding		

Examp	les		
Zamp			
Part B: Plas	tics		
Go to the well	bsite <u>https://www.th</u>	oughtco.com/plastic-chemical-	composition-608930 and answer
the questions	below:		
1. Defin	e <i>plastic</i>		
a.	What is the raw n	naterial for most industrial plast	ics?
b.	List the two types	of plastics	and
c.	What is the prope	rty that <i>plasticity</i> describes?	
2. What	are some different	types of additives in plastics and	d what are some properties that
	affect?	1	1 1
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3. Comp	olete the table:		
		Thermosets	Thermoplastics
Sha	ape		
Str	ucture		

Molecular weight

4. Give three examples of plastics and where you may have encountered them (you may do additional research in other websites if you can't think of three)

5. Properties of plastics depend on three factors. List them.		
6. Define the following:		
a. Monomer		
b. Homopolymer		
c. Copolymer		
7. Describe the following:		
a. Conductivity of plastics		
b. Characteristics of <i>glassy</i> polymers		
c. Rate of degradation of polymers		
Part C: Structures of Polymers		
Go to the website and answer the questions below:		
https://plastics.americanchemistry.com/plastics/The-Basics/		
1. Describe each of the three different types of networks that polymers can form		
2. What type of network is found in		
a. Thermoset polymers		
b. Thermoplastic polymers		
3. Describe the "backbone" of many polymers		

4.	What is the difference between polymers such as polyethylene and polystyrene when compared with polymers such as polyvinyl chloride and Teflon?
5.	Why are polymers such as nylons, polyesters and polycarbonates considered to be inorganic polymers?
6.	Contrast polymers that have an amorphous arrangement with those that have a crystalline arrangement in terms of structure and properties
7.	Give at least two reasons why polymers are used for the following: a. Containers for cleaning products such as Windex b. Coffee cups
	c. Body armor
	d. Two-liter pop bottlese. Kitchen countertops
8.	What is the advantage of making a paper plate from a cellulose material versus the most common way of making polymers?
9.	What is a major challenge in the disposal of waste plastics?

10. Deter	mine the following (from 2005 statistics):
a.	Percentage of trash by weight for plastics
b.	Pounds of polyester recovered from bottles recycled
c.	Pounds of high density polyethylene recovered from bottles recycled
11. What a	are some uses for
a.	Recycled plastics
b.	Plastic from pop and water bottles
c.	Non-recyclable plastics
12. What i	is a problem associated with disposal of plastics in landfills? How is this problem ted?